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18 States of Matter **Three**
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You are already familiar with the three common states of matter: solid, liquid, and gas. Solid objects litter the room around you. For example, you can easily recognize the shape of your desk; you know that your backpack cannot hold seven textbooks. You encounter liquids throughout the day as yo u

Chapter 13: States of Matter

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137 SECTION 13.1 THE NATURE OF GASES (pages 385–389) This section introduces the kinetic theory and describes how it applies to gases. It defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance. Kinetic Theory and a Model for Gases (pages 385–386) 1.

*Name Date Class STATES OF
MATTER 13*

There are three states of matter that we will learn about in this chapter. (If you want to learn about more states of matter, I can refer you to somebody.)

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Answers

Those three states are solid, liquid, and gas. ? These three states are quite different. The main difference is in their particles.

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Chapter 13 States Of Matter

Chapter 13 States of

Matter137 SECTION 13.1 THE

NATURE OF GASES (pages

385-389) This section

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how temperature is related

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Answers and describes how it applies to gases. It defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance.

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Chapter 13: States of Matter. STUDY. PLAY. Kinetic Molecular Theory. Explains the properties of gases in terms of the energy, size, and motion of their particles. Elastic Collision. Describes a collision in which kinetic energy may be transferred between colliding particles but the total kinetic energy of the two particles remains

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Answers

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Chapter 13 - States of

Matter ...

all matter consists of tiny

particles that are

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Answers
constantly in motion What are the three assumptions of the kinetic theory as it applies to gases? -The particles in a gas are considered to be small, hard spheres with an insignificant volume. -The motion of the particles in a gas are rapid, constant, and random.

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Lesson Check; 13.2 The
Nature of Liquids -
Chemistry & You; 13.2 The
Nature of Liquids - 13.2
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Answers
of Matter 2 Kinetic Theory
as Applied to Gases
Fundamental assumptions
about gases. The particles
in a gas are considered to
be small, hard spheres with
an insignificant volume.
Between particles in a gas
there is empty space. No
attractive or repulsive
forces exist between the
particles. 3

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The attacks the USPS
continues to face are not
just attacks on the postal
service but attacks on Black
lives. To defund the USPS
would be to deny future
generations this opportunity

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Answers
and dishonor the legacy of
Black postal workers. Now,
we're taking this matter
into our own hands by
writing and sending
#BlackLoveLetters through
USPS...

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