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Chapter 1 Section

2 MATTER and Its

Properties Matter –

Anything that has
MASS and VOLUME.

ATOMS and

MOLECULES the

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basic building blocks of matter.
ATOM-- the smallest unit of an element that still maintains the properties that element.

ELEMENT- a pure substance made up of only one type of atom.

MATTER and

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Explain why the
flow of a
compressed gas
must be controlled
for practical and
safe use.

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The Periodic Table
and Periodic Law
Section 6.1

Development of
the Modern
Periodic Table 1.
octaves 2. eight 3.
nine 4. accepted 5.
Dmitri Mendeleev
6. atomic mass 7.
elements 8. atomic
number 9. Henry

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Electrons in Atoms

Reviewing

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Vocabulary Match
the definition in
Column A with the
term in Column B.
Column A Column

B 1. The set of frequencies of the electromagnetic waves emitted by the atoms of an element 2. The minimum amount of energy that can be lost or gained

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used in chemistry
can be q... the
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changes (that is,

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energy produced or
absor... A system
that can exchange
neither energy nor
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Change. extensive
property. intensive
property. extensive
properties
(examples)
intensive property
(examples) a
property that
depends on the
amount of matter
in a sample. a
property that

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depends on the
type of matter in a
sample not....
mass, volume,
length, weight.

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number of moles of
known substance C

H C H C H O O O

CO CO CO H O H O

H O number of

moles of unknown
substance O CO H

O C H CO H O C H

O H O C H O CO

use mole ratio 5

mol O 2 1 mol C 3

H 8 3 mol CO 2 1

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mol C₃ H₈ 4 mol
H₂ O₁ mol C₃ H
8 1 mol C₃ H₈ 5
mol O₂

Answers

MATH SKILLS

TRANSPARENCY

MASTER 16

A "chemical change" is a change that produces matter with a different composition than

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the original matter.

Chemical Change A
change in which
one or more

substances are
converted into
different

substances. Heat
and light are often
evidence of a
chemical change.

Properties of
Compounds Quite
different properties

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