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lab this week. *Mole Lab Answers*

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Inc ll iht eered Write the actual mass
here 0.98 x (correct answer) 0.95 x
(correct answer) 0.90 x (correct
answer) 1.02 x (correct answer) 1.05 x
(correct answer) 1.10 x (correct
answer) Correct answer: (actual mass

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divided by 63.546 times 6.022×10^{23}) 4 3
2 Figure 1. Procedure 1.

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(correct answer) $1.10 \times$ (correct
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The fastest way to obtain a relative mass of beans would be to count the beans. The fastest way to obtain a mole of beans would be to weigh them. (At least in principle. The mass of a mole of beans would be incredibly

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large- on the order of 10²² g.) Part III.
All atomic masses agree with the
relative masses to three significant
figures.

*Laboratory Activity 1: Teacher Notes
Continued*

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Have a conceptual understanding of the mole as the method of “counting” items and finding the mass of items that can't be seen. 2. Be able to calculate the number of items (molecules, atoms, ions, and formula

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units) if given the number of moles. 3.
Be able to calculate the number of
moles if given the number of items.

*#20 Introduction to the Mole - Terrific
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target gas law lab worksheet many chemical reactions involve gaseous reactants or products the volumes of gases that react or are produced depend entirely on the moles of the gas and of course on the pressure and

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