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uneventful trans-Atlantic crossing. Suddenly, to the horror of observers on the ground, the airship

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Section 11.1 Assessment. Describe the steps in writing a balanced chemical equation. Write the skeleton equation for the following reactions: Heating solid copper(II)sulfide in the presence of oxygen gas produces pure copper and sulfur dioxide gas. Iron metal and chlorine gas react to form solid iron(III)chloride. $\text{CuS (s)} + \text{O}_2\text{(g)} \rightarrow \text{Cu (s)} + \text{SO}_2\text{(g)}$...

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Section 11.111.1. Objectives. Describe the types of relationships indicated by a balanced chemical equation. State the mole ratios from a balanced chemical equation. Review Vocabulary reactant: the starting substance in a chemical reaction. New Vocabulary stoichiometry mole ratio

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reactions. A few examples of such reactions will be used to develop a clear picture of this classification.

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